

THE NOBEL PRIZE IN CHEMISTRY 2016



Jean-Pierre Sauvage
(Born in 1944, Paris, France)
University of Strasbourg,
France



Sir J. Fraser Stoddart
(Born in 1942, Edinburgh,
United Kingdom)
Northwestern University,
Evanston, IL, USA



Bernard L. Feringa
(Born in 1951, Barger-Compascuum, the Netherlands)
University of Groningen,
the Netherlands

The Nobel Prize in Chemistry 2016 was awarded jointly to **Jean-Pierre Sauvage**, **Sir J. Fraser Stoddart** and **Bernard L. Feringa** "for the design and synthesis of molecular machines".

A tiny lift, artificial muscles and miniscule motors. The Nobel Prize in Chemistry 2016 is awarded to **Jean-Pierre Sauvage**, **Sir J. Fraser Stoddart** and **Bernard L. Feringa** for their design and production of molecular machines. They have developed molecules with controllable movements, which can perform a task when energy is added.

The development of computing demonstrates how the miniaturisation of technology can lead to a revolution. The 2016 Nobel Laureates in Chemistry have miniaturised machines and taken chemistry to a new dimension.

The first step towards a molecular machine was taken by Jean-Pierre Sauvage in 1983, when he succeeded in linking two ring-shaped molecules together to form a chain, called a *catenane*. Normally, molecules are joined by strong covalent bonds in which the atoms share electrons, but in the chain they were instead linked by a freer *mechanical bond*. For a machine to be able to perform a task it must consist of parts that can move relative to each other. The two interlocked rings fulfilled exactly this requirement.

The second step was taken by Fraser Stoddart in 1991, when he developed a *rotaxane*. He threaded a molecular ring onto a thin molecular axle and demonstrated that the ring was able to move along the axle. Among his developments based on rotaxanes are a molecular lift, a molecular muscle and a molecule-based computer chip.

Bernard Feringa was the first person to develop a molecular motor; in 1999 he got a molecular rotor blade to spin continually in the same direction. Using molecular motors, he has rotated a glass cylinder that is 10,000 times bigger than the motor and also designed a nanocar.

2016's Nobel Laureates in Chemistry have taken molecular systems out of equilibrium's stalemate and into energy-filled states in which their movements can be controlled. In terms of development, the molecular motor is at the same stage as the electric motor was in the 1830s, when scientists displayed various spinning cranks and wheels, unaware that they would lead to electric trains, washing machines, fans and food processors. Molecular machines will most likely be used in the development of things such as new materials, sensors and energy storage systems.

http://www.nobelprize.org/nobel_prizes/chemistry/laureates/2016/